

The impact of dust on interstellar gas



NWO
Nederlandse Organisatie voor Wetenschappelijk Onderzoek

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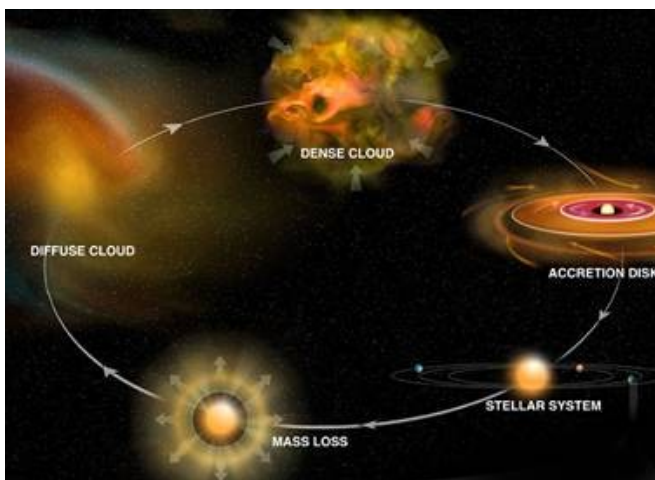
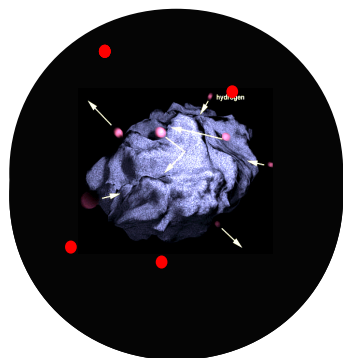
The low-metallicity ISM: chemistry, turbulence and magnetic fields

Overview

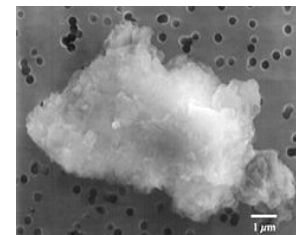
Star form in clouds made of gas + dust

Catalyst: Enrich gas

H_2 , H_2O , O_2 , H_2O_2



Reservoir:
Stealing the gas
Formation of ices

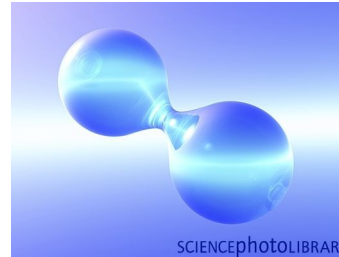


Impact of dust on the gas? Impact of dust on star formation?

Low metallicity? Traces of dust affect the chemistry?

Formation of H₂

Gas phase route:



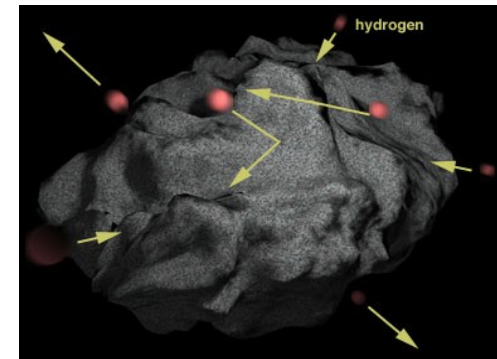
Grain surface route:

H₂ in MW **NOT** explained by gas phase reactions :

Grain *Gould & Salpeter 1963*

H₂ formation dust >> gas

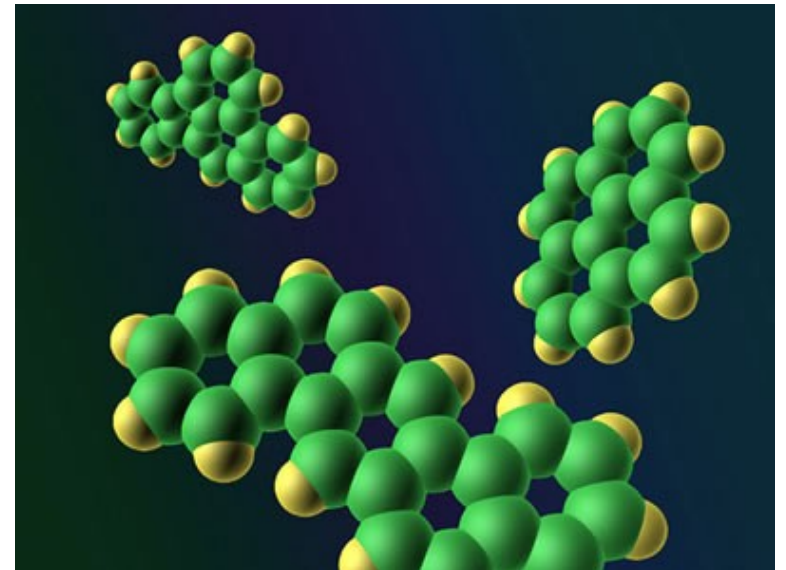
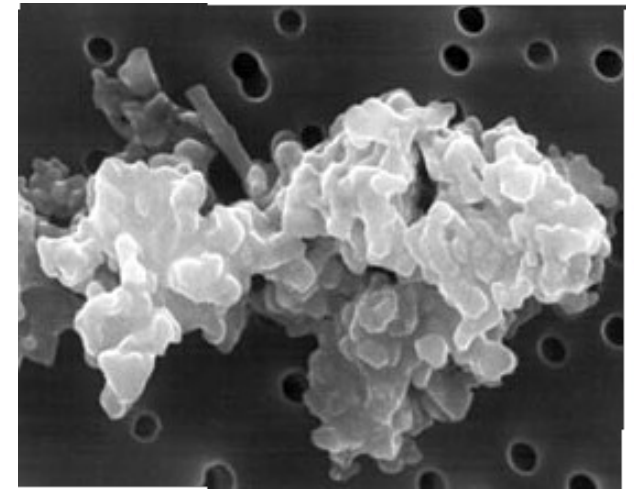
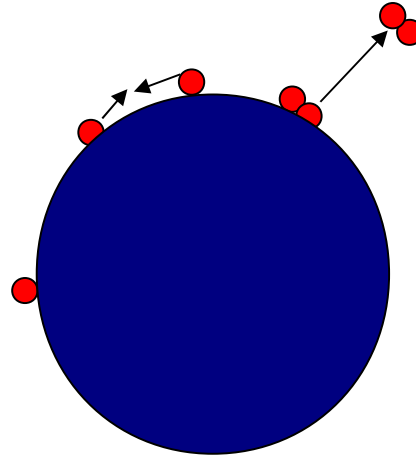
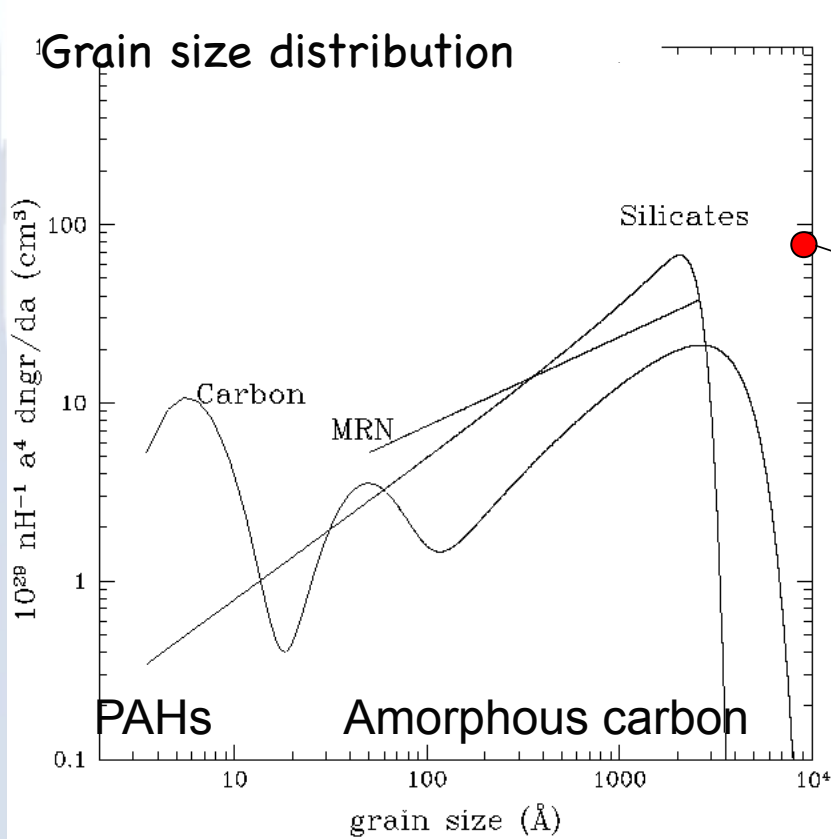
Reaction exothermic → products gas



H₂ formation dust VS gas with metallicity?

Are traces of dust enough to form H₂?

Interstellar dust grains



Weingartner & Draine 2001
Mathis, Rumpl & Nordsieck 1977

DUST= Silicates, Amorphous carbon, PAHs
PAHs = 50% of surface available for chemistry

Formation of H_2 on dust

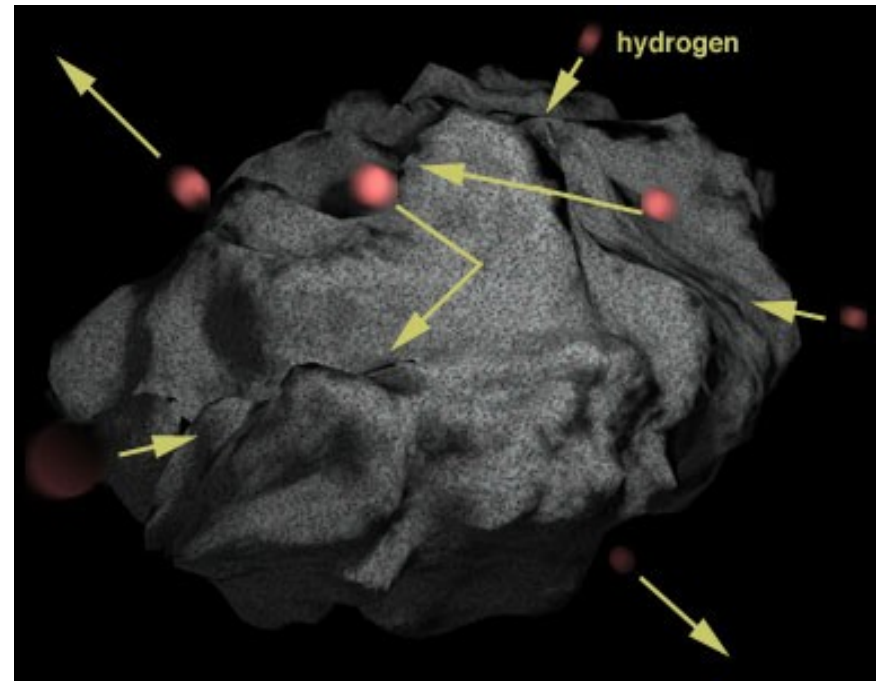
1) Interactions atom/surface
Experiments: TPD

Ab-initio calculations

2) Mobility on the surface

Simulations \rightarrow efficiency H_2 formation

Early Universe. For which Z_0 dust boosts the formation of H_2 ?

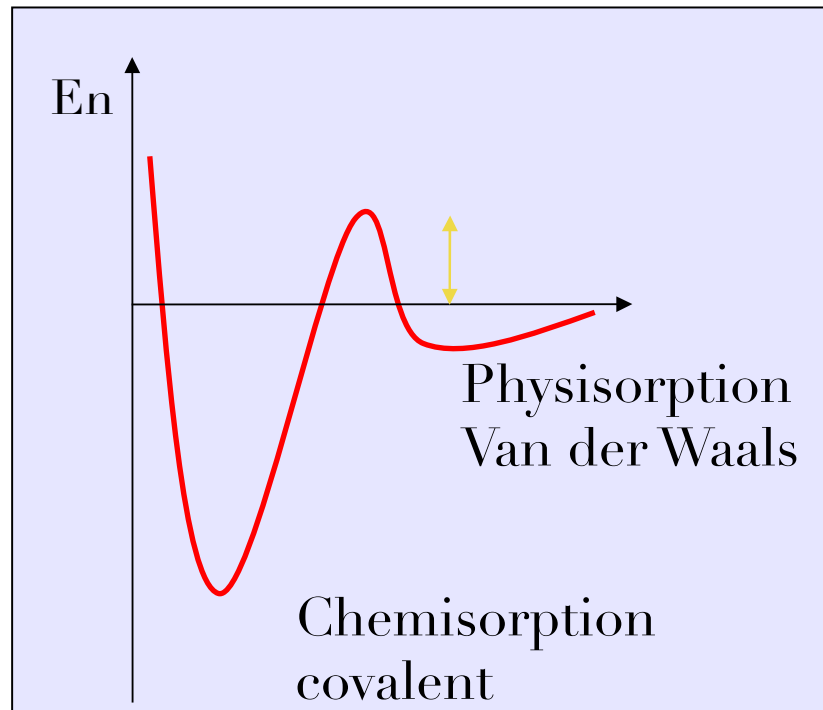
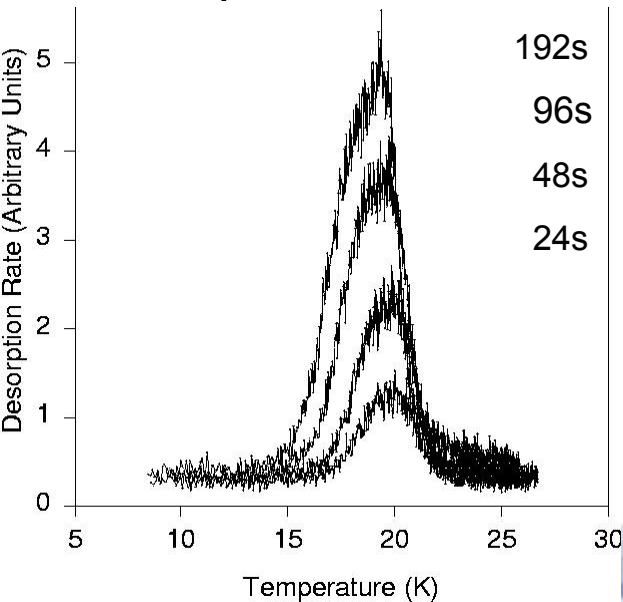


Interaction atom/surface: experiments

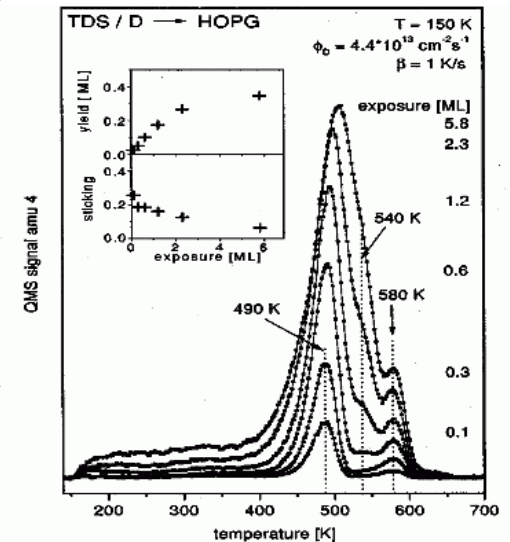
Experiments on graphite, amorphous carbon, silicates

Pirronello et al. 1997, 1999, Zecho et al. 2002, Perets et al. 2007, Vidali et al. 2007

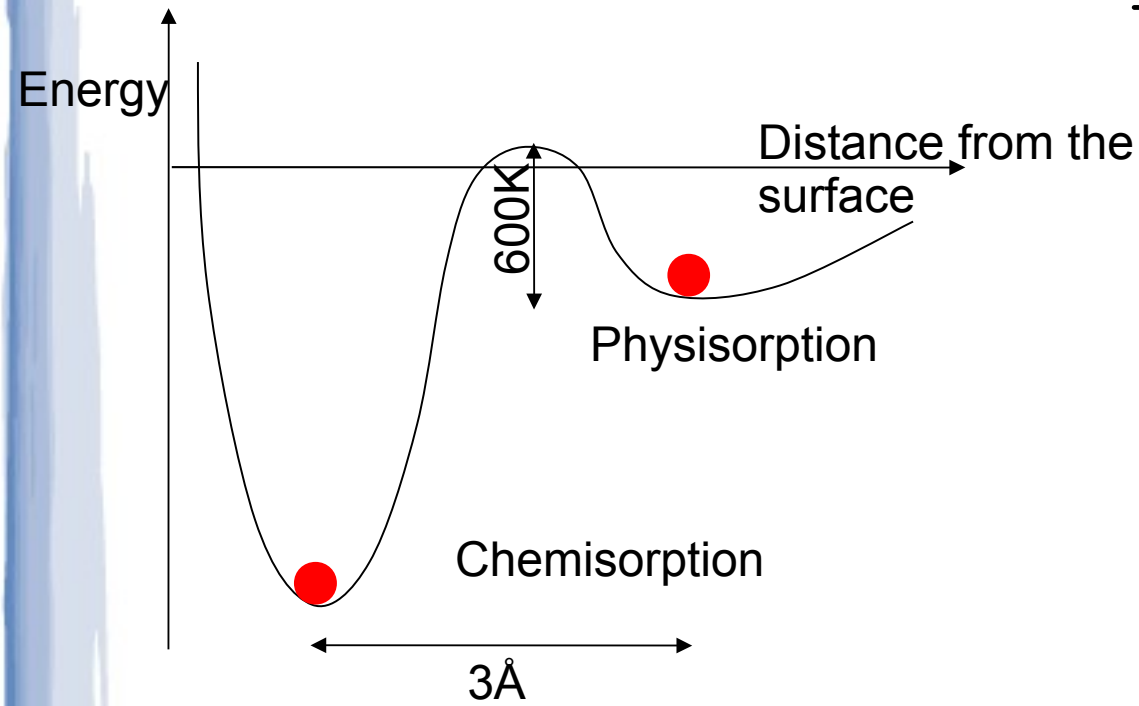
Amorphous Carbon
Low temp., $T_{surf} = 5\text{ K}$



Graphite
High temp., $T_{surf} = 150\text{ K}$

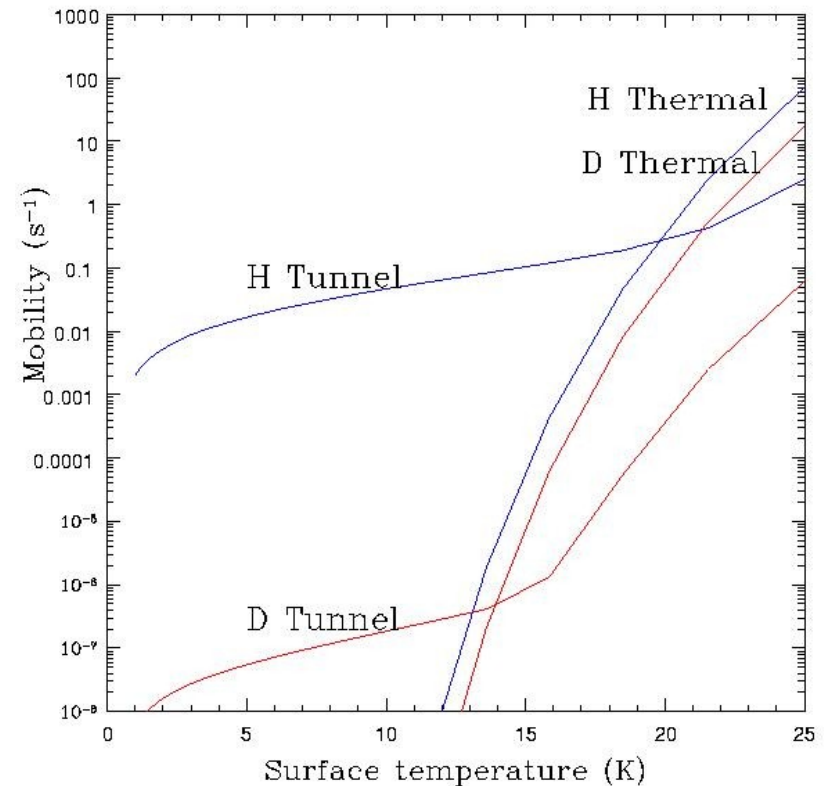


Model: Interaction and mobility

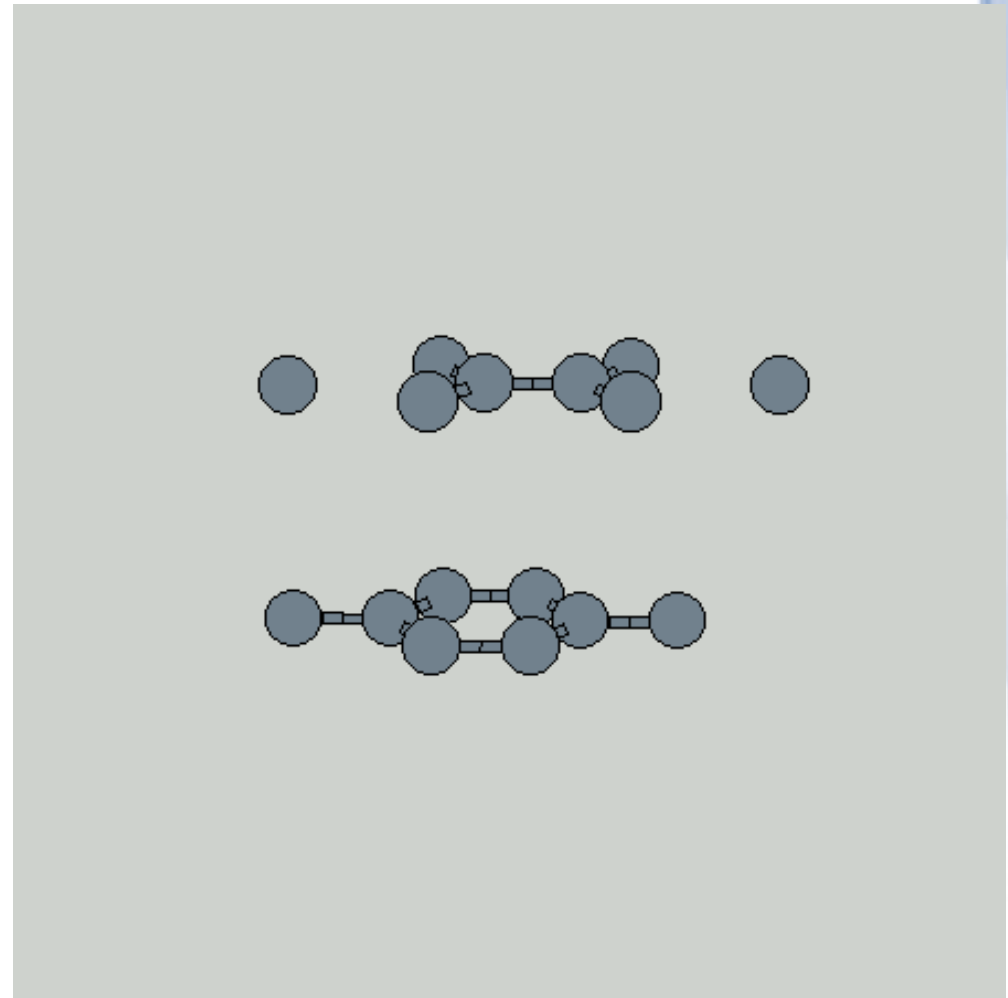
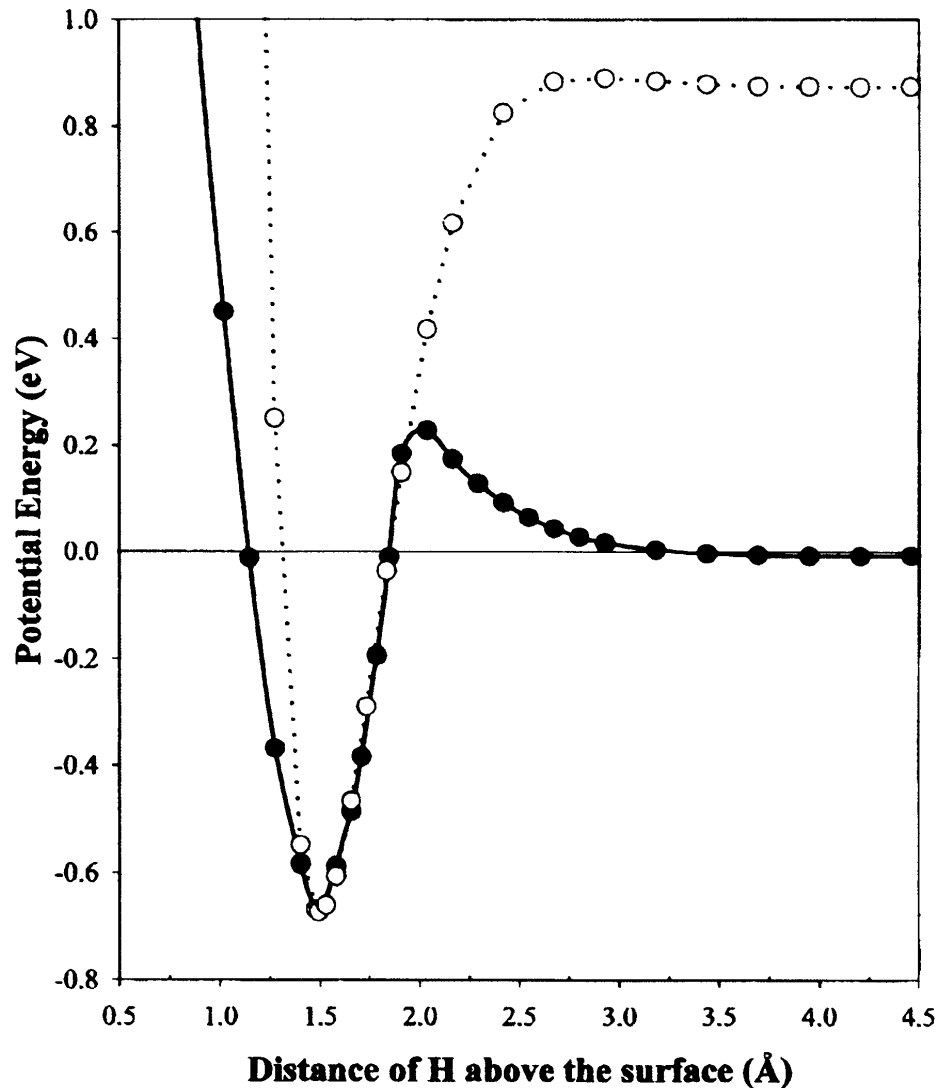


- Physisorption + chemisorption
- tunnel + thermal hopping

Transmission coefficient of the barriers
→ mobility of H and D atoms

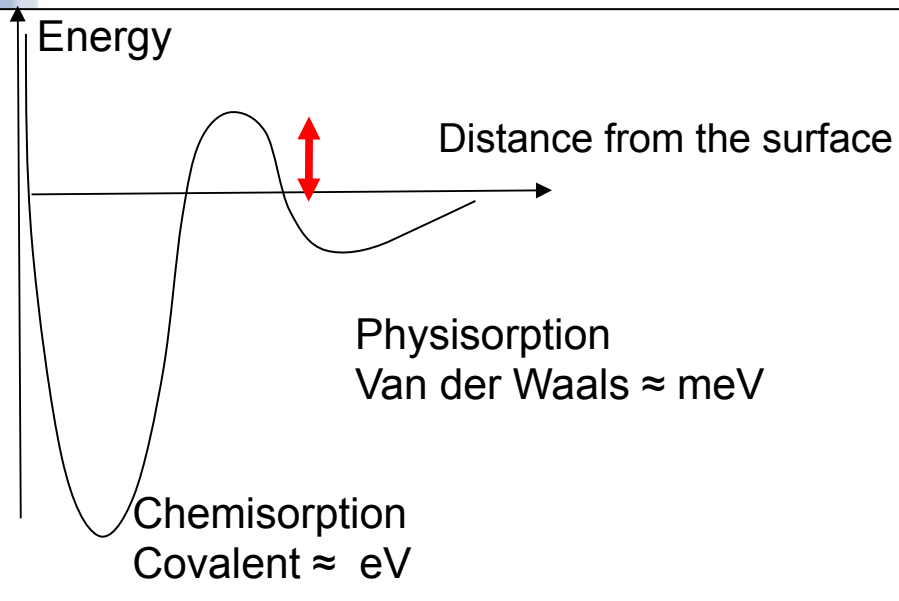


Interaction atom/surface: Density functional theory (DFT)



Eva Rauls

Interaction atom/surface: experiment



Graphite:

Chemisorption of H
C puckered out of the basal plane
associated with barrier \sim 0.2 eV.

Jeloaica & Sidis 1999

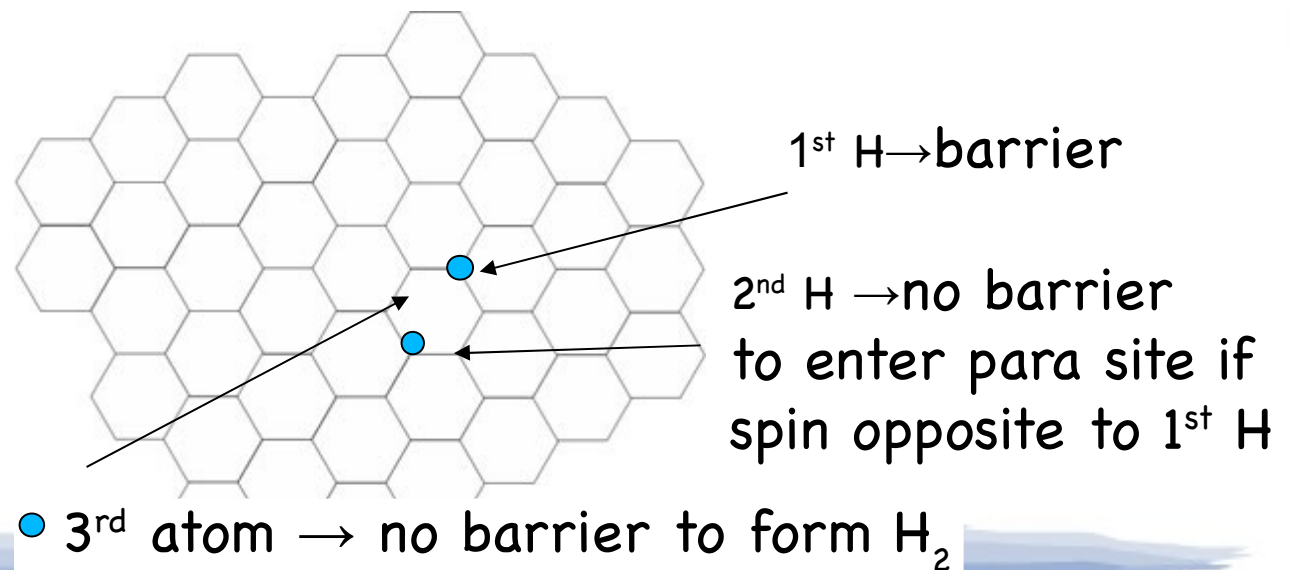
Sha & Jackson 2002

Recent studies:

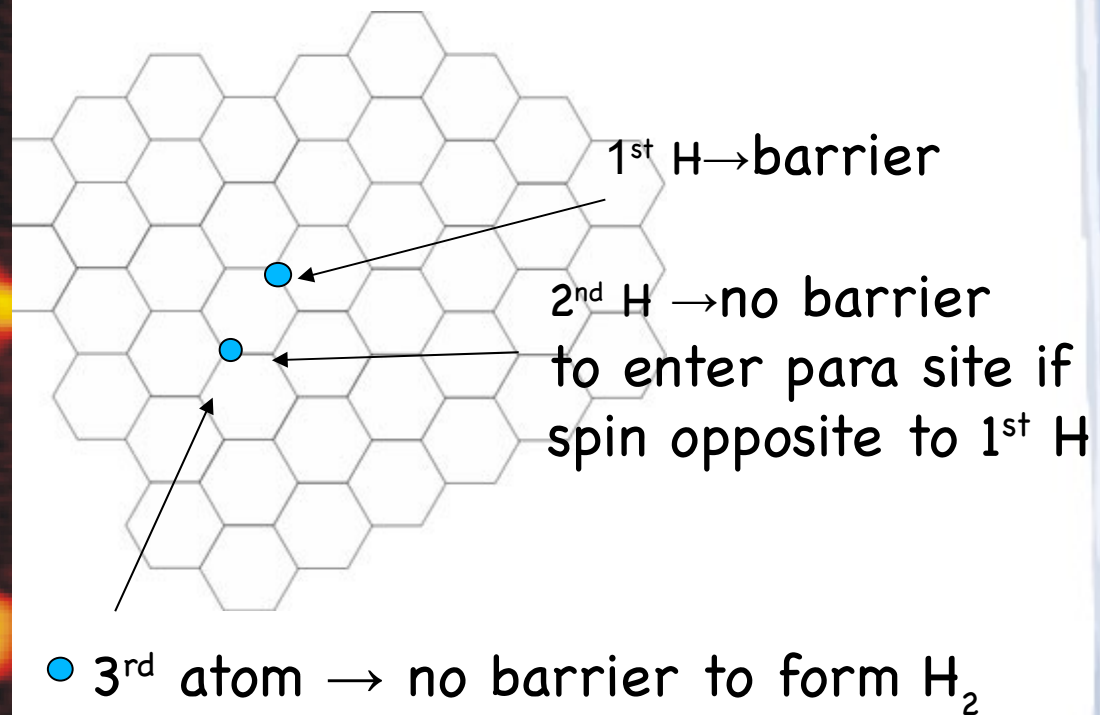
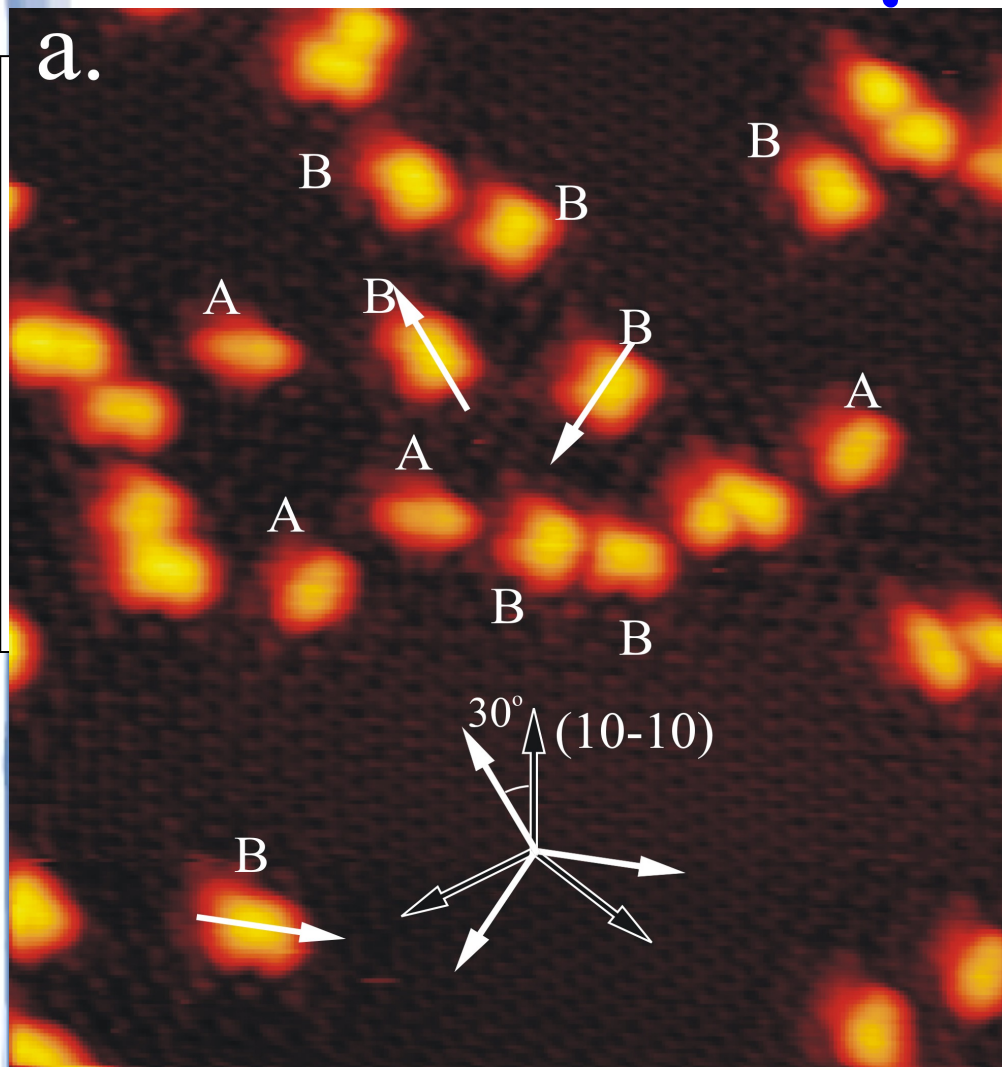
Hoernekaer et al. 2006

Rougeau et al. 2006

Bachelier et al. 2007



Interaction atom/surface: experiment

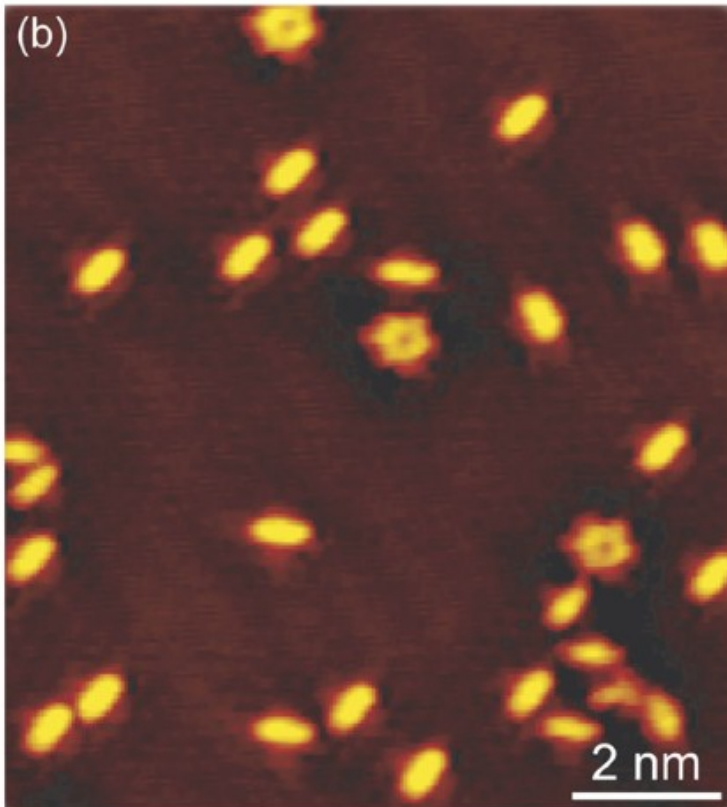


STM @ 170K

Hornekaer 2006

Interaction atom/surface: experiment

- 1 atom sticks \rightarrow dust becomes catalyst \rightarrow H₂ formation barrier-less



Atoms get grouped as

Dimers (2 atoms)

Trimers (3)

Hexamers (6)

Binding energy increases with number of atoms

Grain surface chemistry: Monte carlo simulations

Atoms arrive randomly from gas phase

$$\text{Flux} = n_x v_x \sigma \text{ (s}^{-1}\text{)}$$

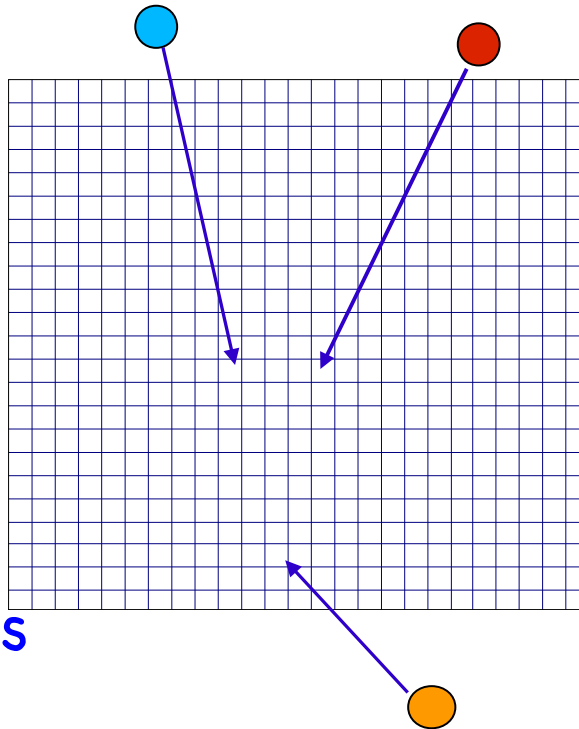
On the grid

random walk

UV + CR

Evaporation

Formation of molecules

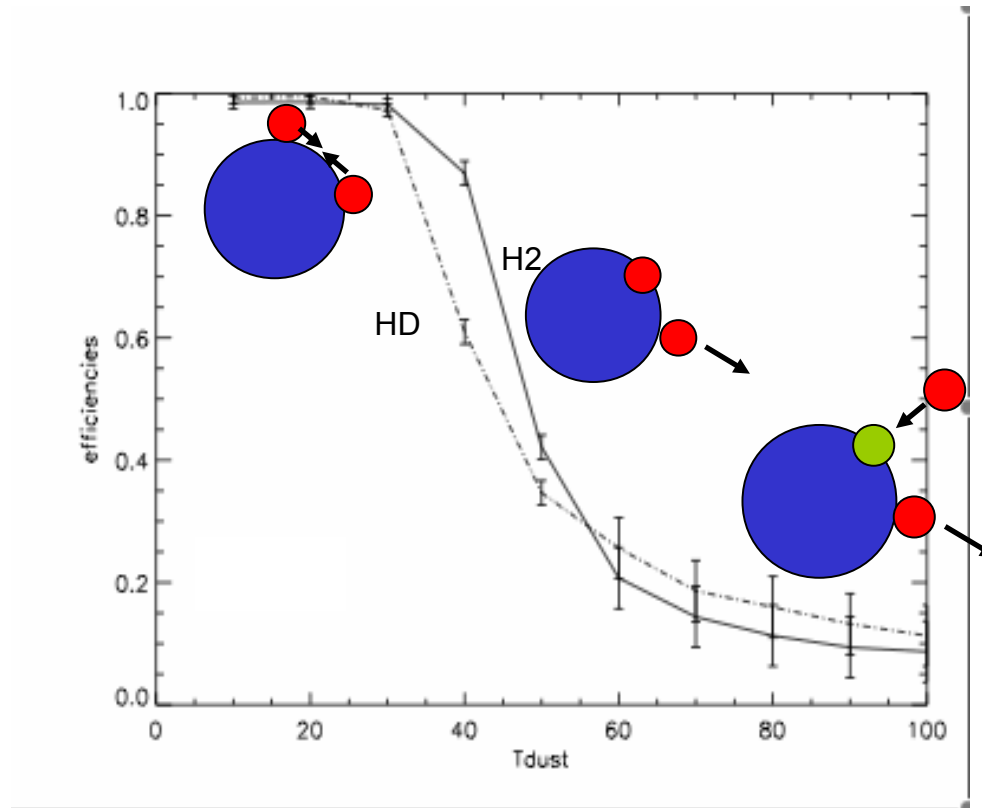


Grain surface = grid

Each point of the grid:

site atom/molecule

Formation of H₂ on dust



Formation of H₂ and HD

physisorbed atoms @ low T_{dust}

chemisorbed atoms @ high T_{dust}

clusters of atoms \rightarrow H₂ and HD high T_{dust}

H_2 and HD in the early Universe

First stars cooled by H_2 → very massive $\sim 100M_{\odot}$

Stars cooled by HD few $\sim 10M_{\odot}$

Amount of coolant available → essential to star formation

H_2 and HD at low Z_{\odot} **dust VS gas?**

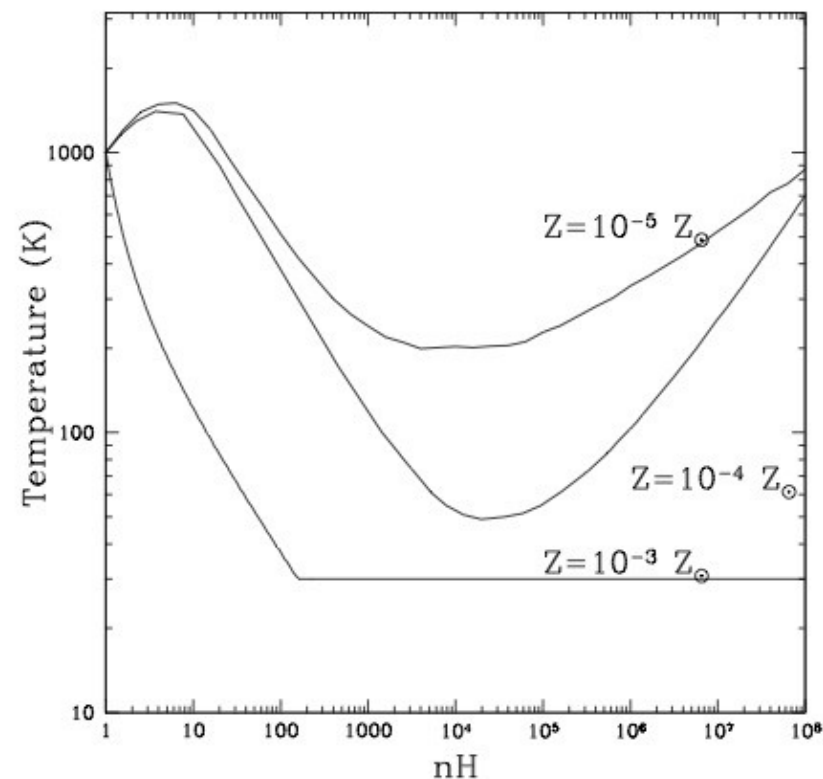
→ compute H_2 and HD formation during cloud collapse.

H₂ and HD in the early Universe

Cloud collapse with 1 atom cm^{-3} @ $z=10$

Temperatures profiles depend on

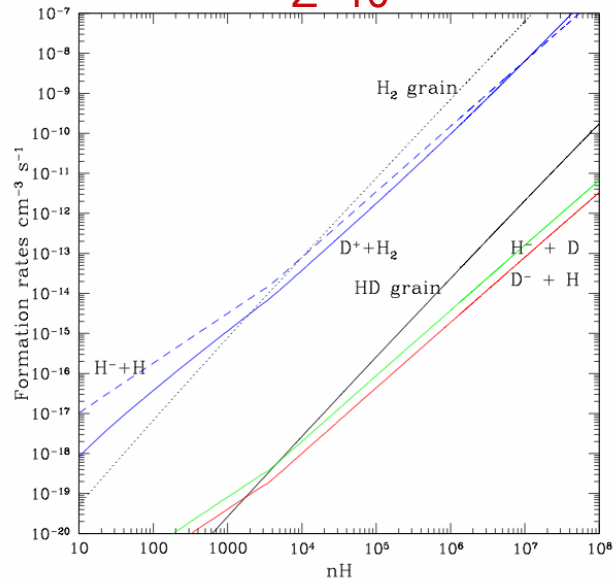
- Adiabatic heating
- Cooling by H₂ and HD (when no metals, Glover & Abel 2000)
- Cooling by fine structure lines (Meijerink & Spaans 2005)



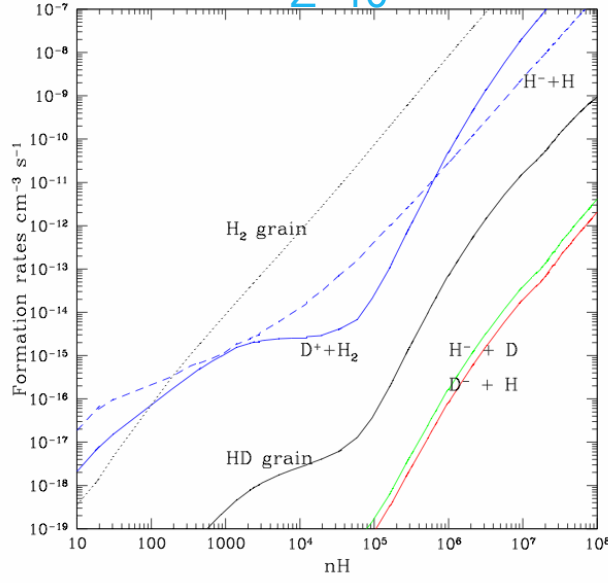
H₂ and HD in the early Universe

Cazaux & Spaans 2009

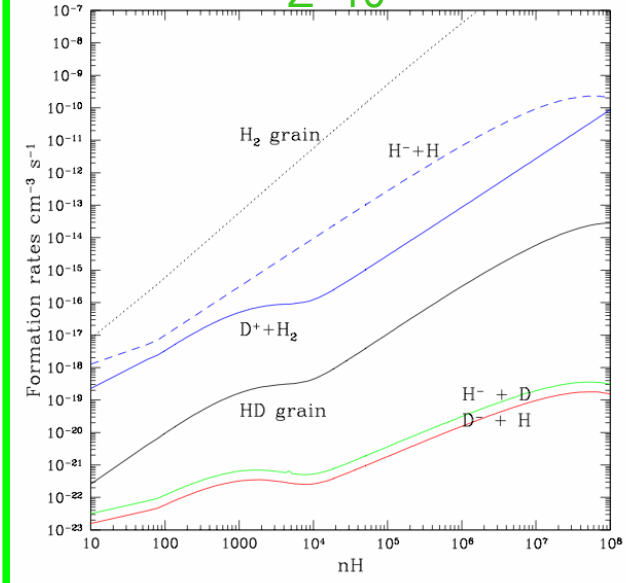
Z=10⁻⁵



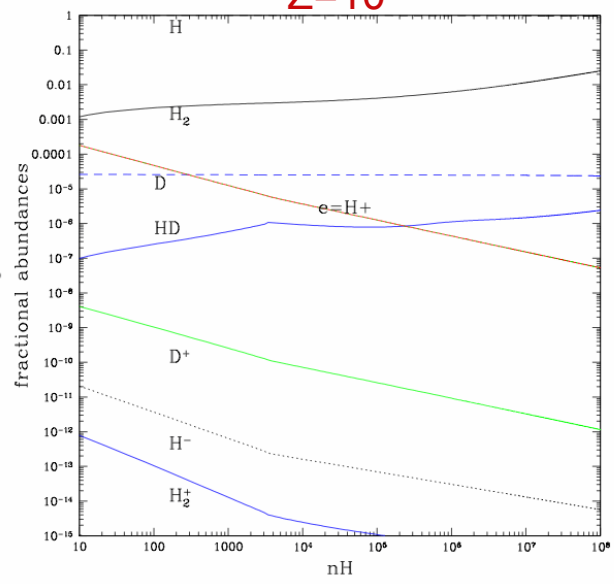
Z=10⁻⁴



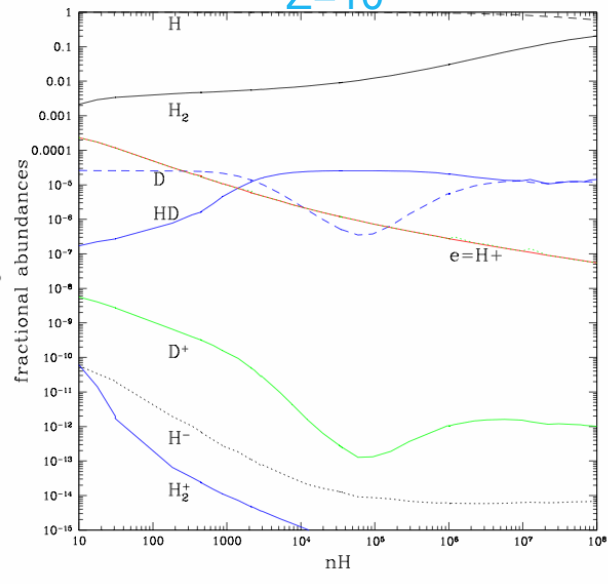
Z=10⁻³



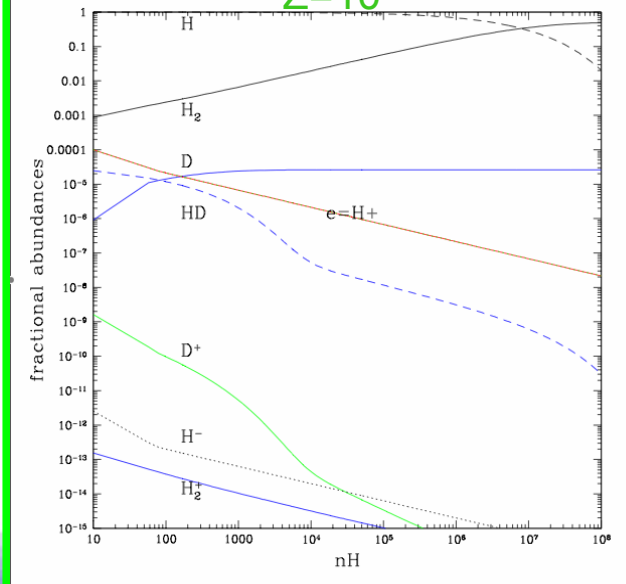
Z=10⁻⁵



Z=10⁻⁴



Z=10⁻³



H₂ and HD in the early Universe: Conclusions

- Traces of dust → H₂ on dust grain most efficient route.
- H₂ produced → HD forms through $D^+ + H_2 \rightarrow$ Grain surface routes never dominate.
- Traces of dust boost H₂ which boost HD →
These coolants could impact star formation (Work in progress).

Formation of molecules on dust

Water is an important coolant of warm dense clouds (Neufeld 1995)

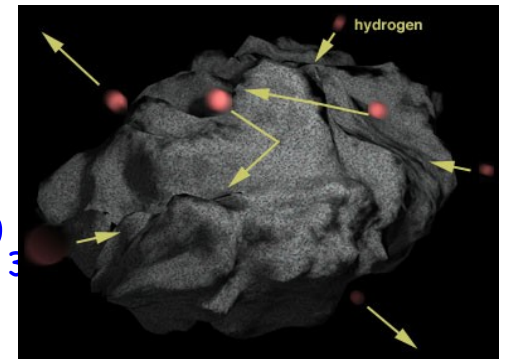
Observed by Herschel in XDR

H_2O forms \rightarrow **gas**

- In warm environments $\text{H}_2 + \text{O}$
- Cold shielded cores $\text{H}_3^+ / \text{H}_3\text{O}^+$

\rightarrow **dust**

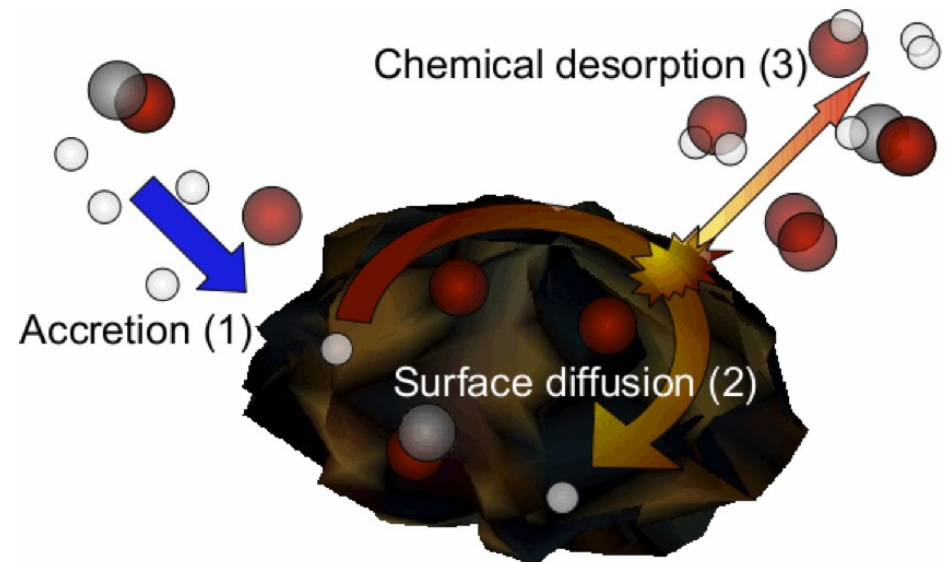
- Several routes involving O , O_2 and O_3



Formation of molecules on dust

Exothermic reaction \rightarrow product released in gas.

$\text{H} + \text{H} \rightarrow \text{H}_2$	60 %
$\text{H} + \text{O} \rightarrow \text{OH}$	30 %
$\text{OH} + \text{H} \rightarrow \text{H}_2\text{O}$	90 %
$\text{H}_2 + \text{O} \rightarrow \text{OH} + \text{H}$	0 %
$\text{H}_2 + \text{OH} \rightarrow \text{H}_2\text{O} + \text{H}$	0 %
$\text{O} + \text{O} \rightarrow \text{O}_2$	60 %
$\text{O} + \text{O}_2 \rightarrow \text{O}_3$	50 %
$\text{H} + \text{CO} \rightarrow \text{HCO}$	80 %



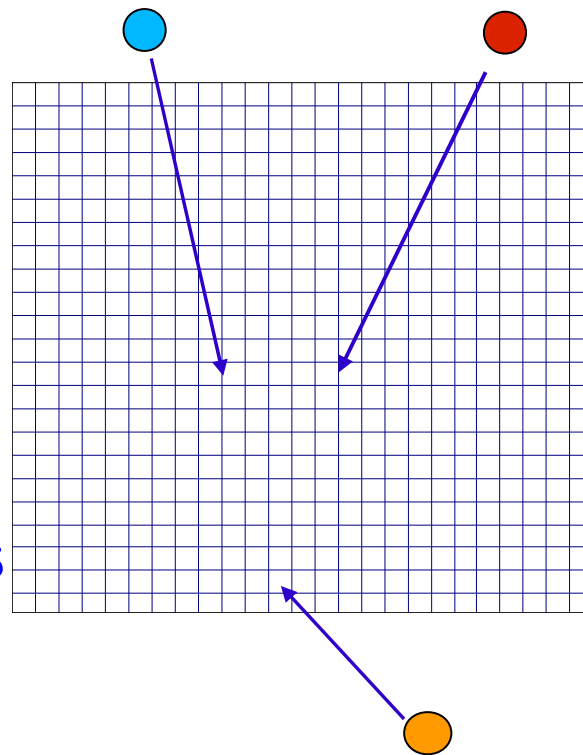
Derived from exp (Dulieu et al. 2012)

Formation of molecules on dust: Monte carlo simulations

Atoms arrive randomly from gas phase

$$\text{Flux} = n_x v_x \sigma \text{ (s}^{-1}\text{)}$$

- On the grid
- random walk
- UV + CR
- Evaporation
- Formation of molecules



Grain surface = grid
Each point of the grid:
site atom/molecule

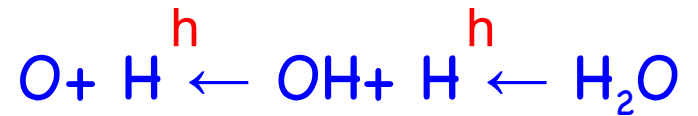
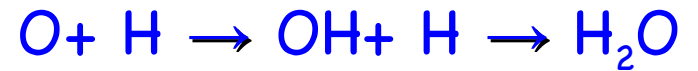
$\text{H}_2, \text{HD}, \text{D}_2, \text{OH}, \text{OD}, \text{O}_2, \text{H}_2\text{O}, \text{HDO}, \text{D}_2\text{O}, \text{O}_3, \text{HO}_2, \text{DO}_2, \text{H}_2\text{O}_2, \text{HDO}_2, \text{D}_2\text{O}_2$

Diffuse clouds

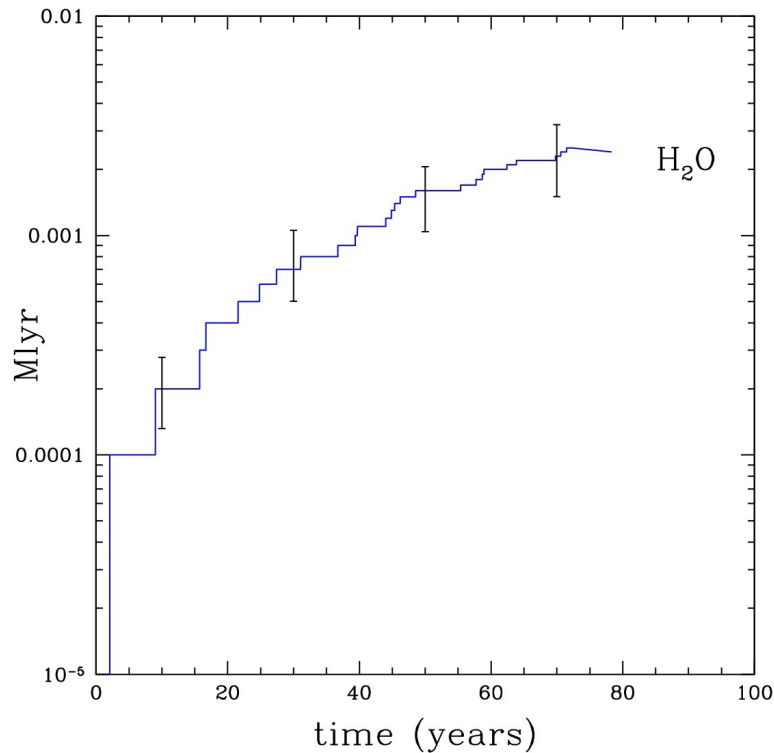
Diffuse clouds: H atomic

$T_{\text{dust}}=18\text{K}$, $T_{\text{gas}}=100\text{K}$

$n\text{H}=100\text{ cm}^{-3}$, $\text{O}/\text{H}=3 \cdot 10^{-4}$, $D/\text{H}=2 \cdot 10^{-5}$



Grain surface



Gas phase

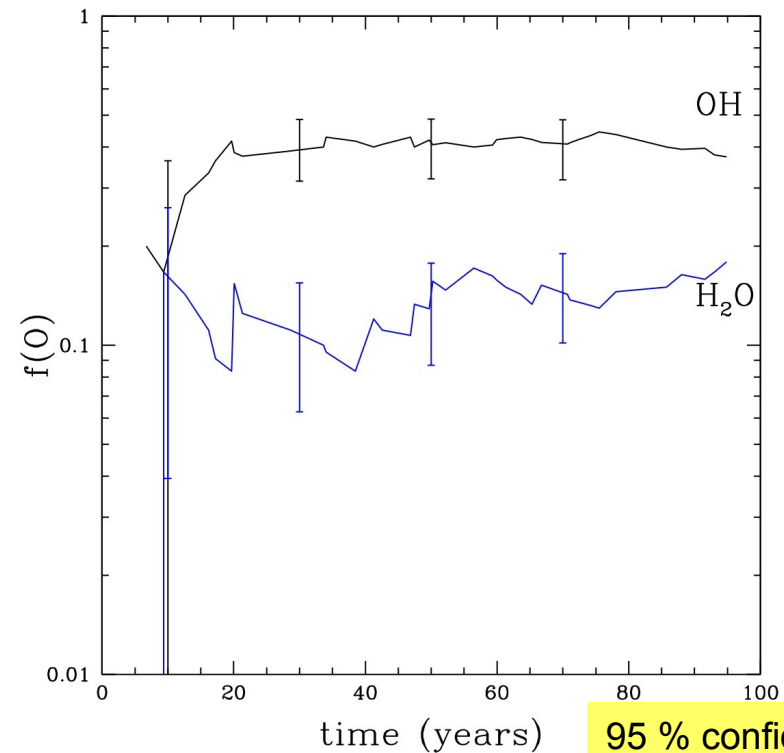


Photo-dissociation regions

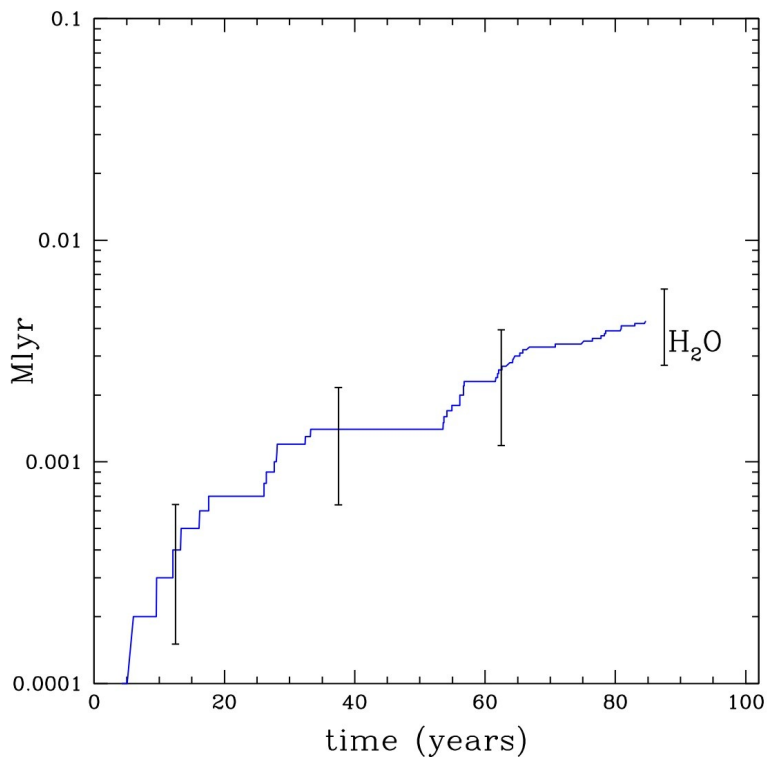
PDR: H molecular

$T_{\text{dust}}=30\text{K}$, $T_{\text{gas}}=30\text{K}$, $G_0=10^3$, $A_V=5$

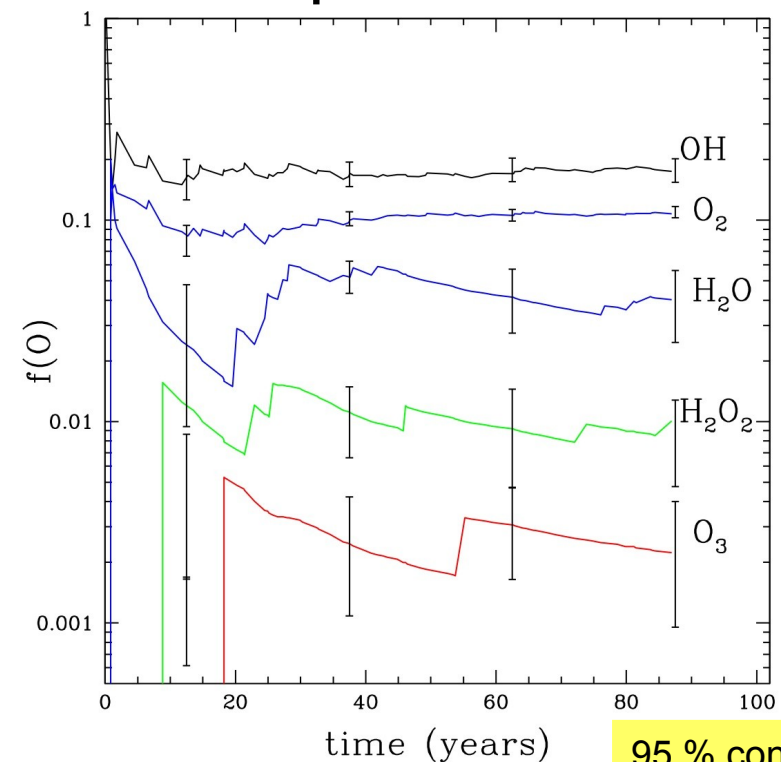
$n_{\text{H}}=1000\text{ cm}^{-3}$, $\text{O}/\text{H}=3 \cdot 10^{-4}$, $D/\text{H}=2 \cdot 10^{-5}$

H_2O forms with O_2 and O_3

Grain surface



Gas phase



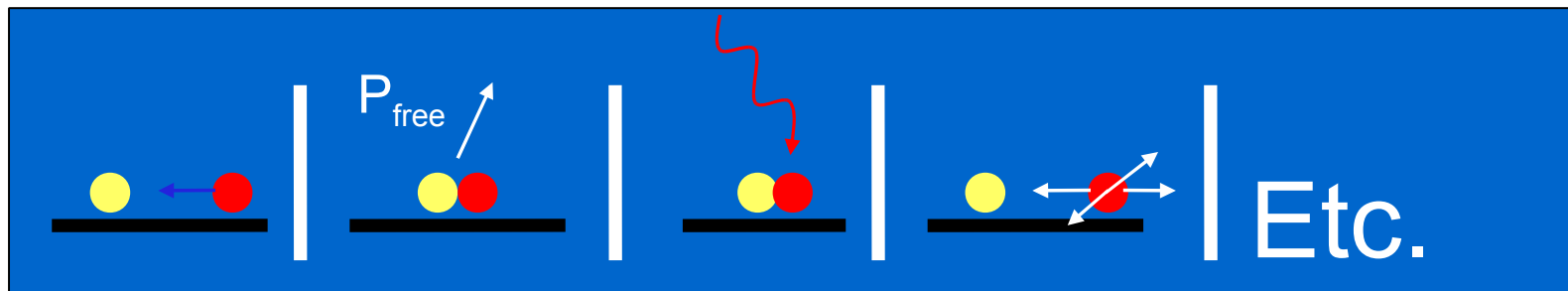
95 % confidence level

Formation of molecules on dust

Cold grains (~ 10 K) favours **hydrogenation**

Warmer grains (30 K) favours **oxygenation**

UV photons dissociate species that **recombine**. "dissociation-formation-dissociation" boost gas phase.



Species released in gas \rightarrow photo-dissociated.
Boost VS photo-dissociation?

Star formation

How does dust (and metallicity) impact SF?

MHD simulations: Flash

gas phase + dust chemistry

Star formation

Initial cloud conditions: 10^3 cm^{-3}

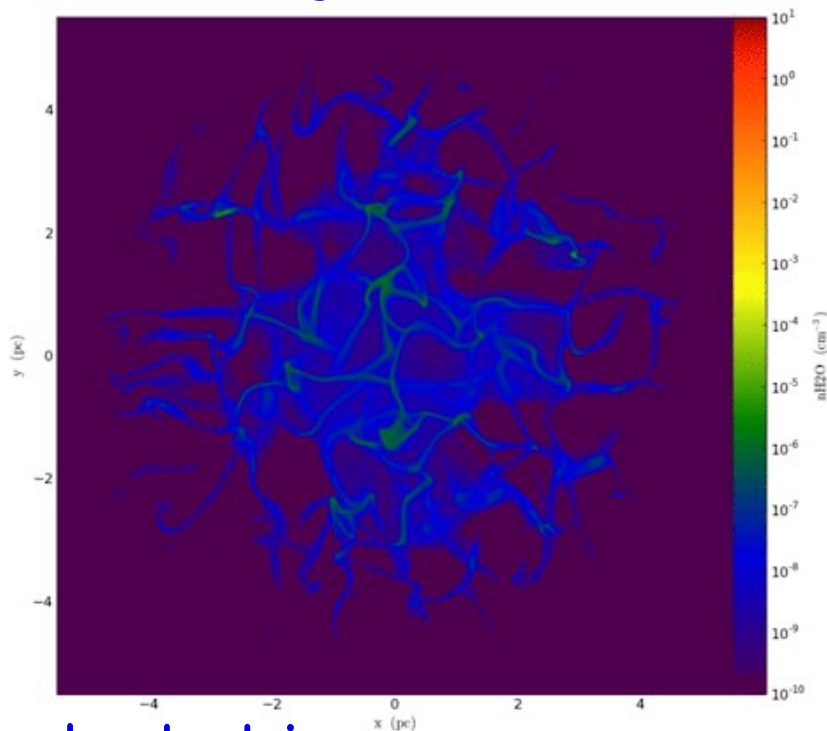
Isothermal: $T_{\text{gas}} = T_{\text{dust}} = 20\text{K}$

10 chemical species

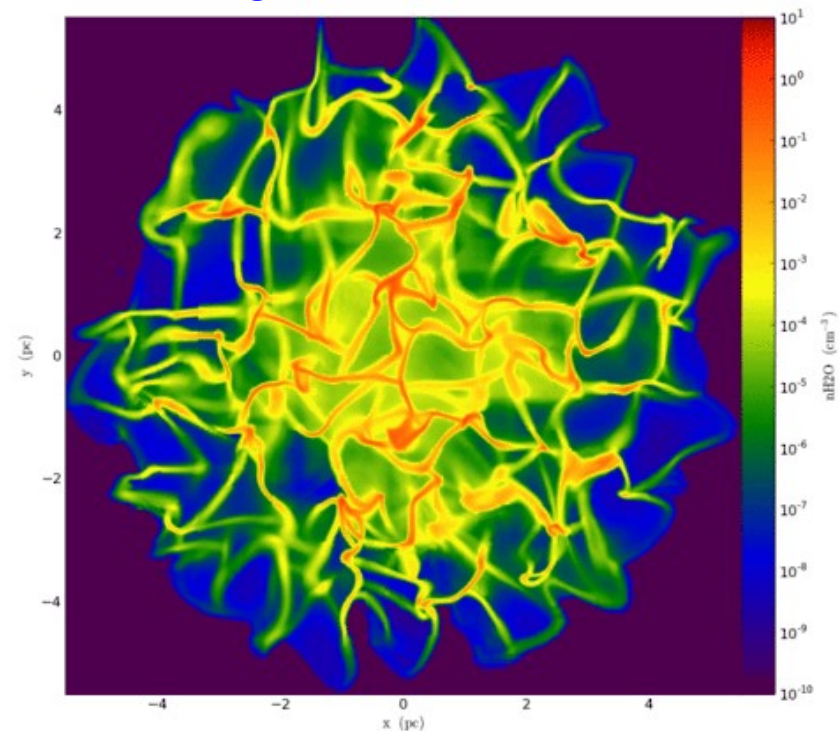
Gas phase: 40 reactions

Dust: H_2 and H_2O

gas only



gas + dust

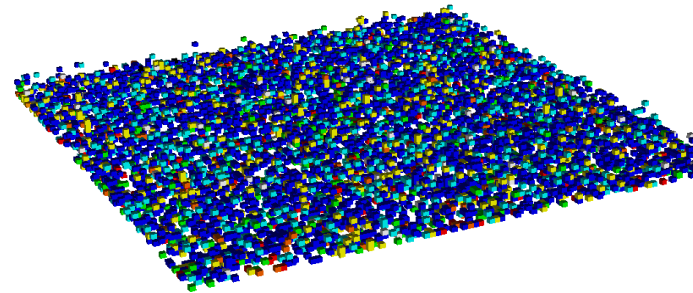
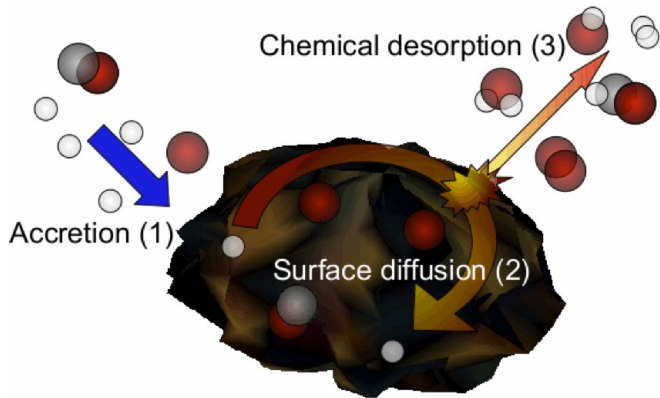


Hocuk et al in prep.

Star formation

Include cooling and heating mechanisms

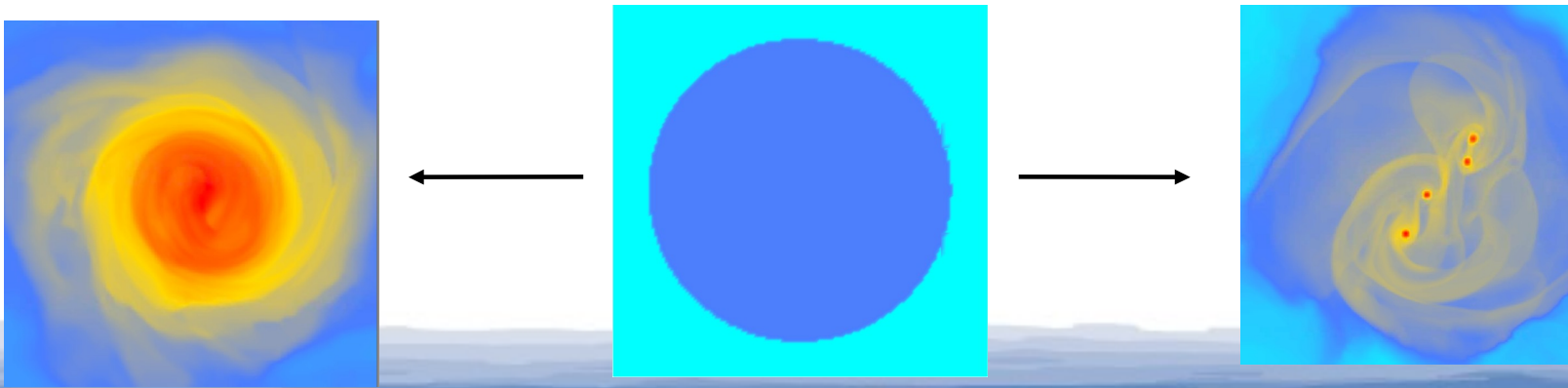
Extend the chemical network (dust + gas)



H₂
H
O
H₂O
CO
H₂CO

Fragmentation of Molecular cloud

Dust impact fragmentation/ SF efficiency and IMF.



Summary and Conclusions

H₂ formation on dust involves:

H weakly bound to the surface.

H strongly bound to the surface and making pairs (or groups).

Traces of dust ($10^{-5} Z_{\odot}$) are enough to form H₂.

H₂ forms on dust → boost HD through $D^+ + H_2$

Other molecules forming on dust → release in the gas if exothermic reactions → direct impact on the gas (coolants).

Hydrodynamic simulations + dust / gas chemistry + cooling → Star formation varies with Z_{\odot} (scales with dust).